In all things there is a law of cycles. - Publius Cornelius Tacitus

TOPICS INCLUDE: • Biodiversity

• Biogeochemical Cycles

AP ENVIRONMENTAL SCIENCE



UNIT 2: THE LIVING WORLD (PART B)

II. THE LIVING WORLD (10-15%)

A. Ecosystem Structure – biological populations and communities, ecological niches, interactions among species, keystone species, species diversity and edge effects, major terrestrial and aquatic biomes

B. Energy Flow – photosynthesis and cellular

- respiration, food webs and trophic levels and ecological pyramids
- C. Ecosystem Diversity biodiversity, natural selection, evolution, and ecosystem

services D. Natural Ecosystem Change – climate

- shifts, species movement and ecological succession
- E. Natural Biogeochemical Cycles –
- carbon, nitrogen, phosphorus, sulfur, water, and conservation of matter

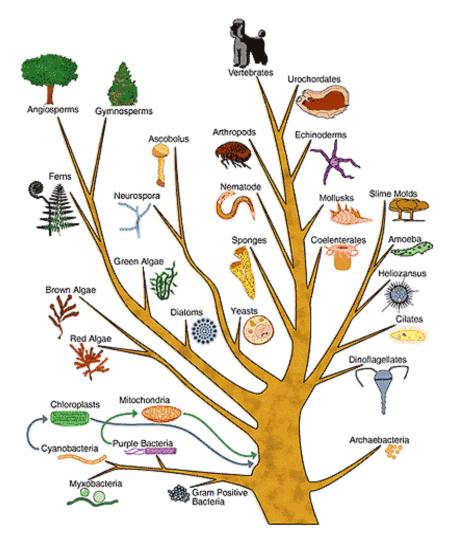


ECOSYSTEM DIVERSITY

 <u>biodiversity</u>: number & variety of organisms found within an ecosystem

DIVERSITY INCREASERS	DIVERSITY DECREASERS
Diverse habitats	Environmental stress
Disturbance in the habitat (e.g. fire, storms)	Extreme environments
Environmental conditions w/low variation	Extreme limitations in the supply of a fundamental resource
Trophic levels w/high diversity	Extreme amounts of disturbance
Middle states of succession	Introduction of species from other areas
Evolution	Geographic isolation

EVOLUTION

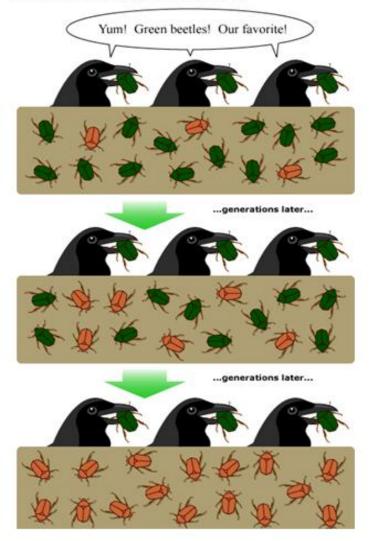


• change in the genetic composition of a population during successive generations as a result of natural selection acting on the genetic variation among individuals and resulting in the development of new species

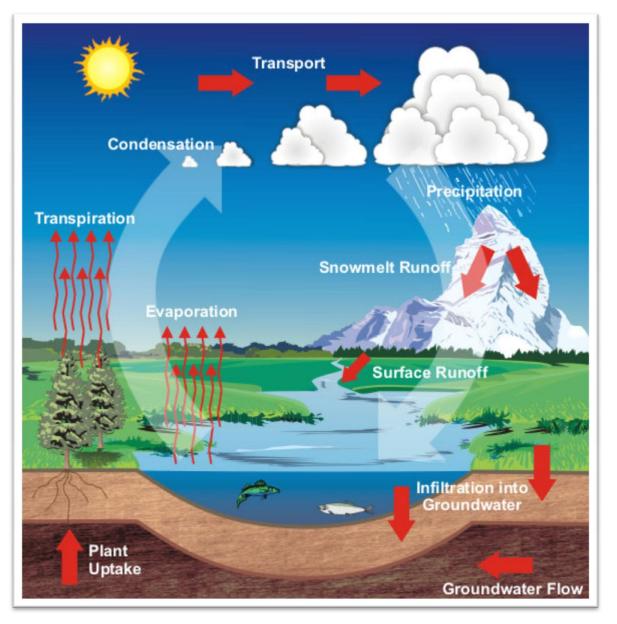
NATURAL SELECTION

- mechanism how organisms evolve
- works on the individual by determining which individuals have adaptations that allow them to survive, reproduce and pass on those adaptive traits to their offspring
- survival of the "fittest"; fittest means ability to reproduce and pass on genes to offspring

Natural selection, in a nutshell:



WATER CYCLE



CARBON CYCLE

- key events: photosynthesis
 & respiration (living things are exchange pools for carbon)
- organisms die get buried and subjected to extreme heat & pressure →organic matter converts to oil, gas and coal (fossil fuels)
- carbon reservoirs: oceans (CO₂ soluble in water) & earth's rocks (calcium carbonate)

The Global Carbon Cycle Atmosphere Land-use Photosynthesis Changes & Respiration **Fossil Fuel** Occan-Atmosphere Combustion Exchange & Cement Manufacture Surface errestria Ocean Photosynthesis River Runoff & Respiration Lithosphere Phytoplankton Sinking Ocean Particles Circulation

NITROGEN CYCLE

• STEP 1: NITROGEN FIXATION

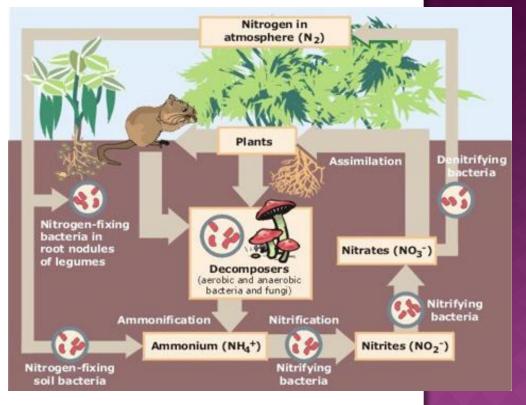
- → Nitrogen must be converted (fixed) into ammonia (NH₃) or nitrates (NO₃⁻) to be used by living organisms
- → Occurs as result of *Rhizobium* (soil bacteria) found in roots of legumes (beans/clover) and/or atmospheric effects (lightning)

• STEP 2: NITRIFICATION

→ Soil bacteria converts ammonium ions (NH₄⁺) into one of the forms used by plants; nitrate (NO₃)

• STEP 3: ASSIMILATION

→ plants absorb ammonium (NH₃) ammonium ions (NH₄⁺) & nitrate ions (NO₃⁻) through their roots (heterotrophs obtain N by consuming plants)



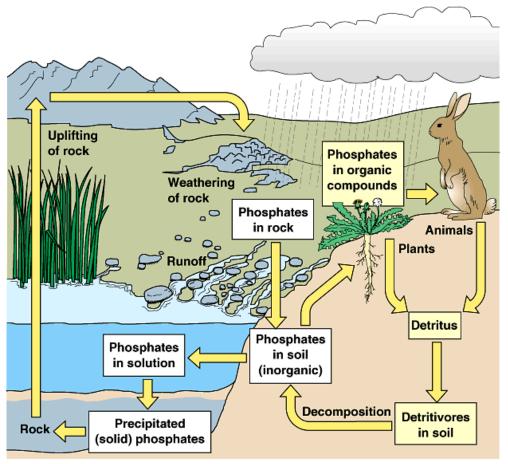
• STEP 4: AMMONIFICATION

→ decomposing bacteria convert dead organisms & other waste to ammonia (NH₃) or ammonium ions (NH₄⁺) (reused by plants

• STEP 5: DENITRIFICATION

→ specialized bacteria convert ammonia to NO₃⁻, NO₂, and N₂ (release it back to atmosphere)

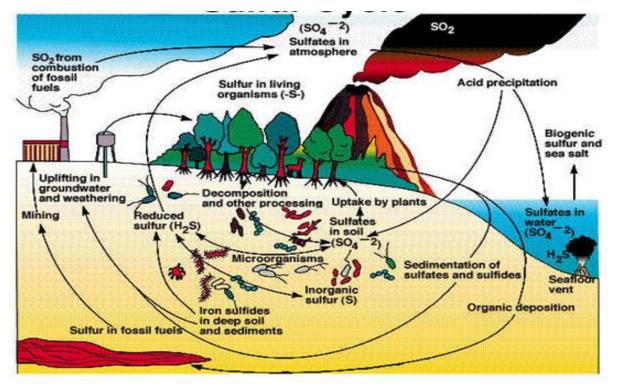
PHOSPHORUS CYCLE



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- only exists in atmosphere as dust particles
- P is major component of nucleic acids
- found in soil, rock & sediments (released through weathering)
- released in the form of phosphate (PO₄³⁻); soluble and absorbed though soil by plants
- often limiting factor in plant growth

SULFUR CYCLE



- makes up proteins and vitamins
- plants absorb S when dissolved in water; animals obtain it by consuming plants
- most found in rocks & salts or buried in ocean sediment
- enters atmosphere through volcanic eruptions, bacterial functions, decay of organisms, human activity